

Effect of post-deformation annealing on the R-phase transformation temperatures in NiTi shape memory alloys

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Abstract

In NiTi shape memory alloys, both the annihilation of dislocations and the formation of Ni_4Ti_3 precipitates may occur during post-deformation annealing. Different responses of the R-phase transformation temperatures to the annealing conditions have been reported. In order to find out the main factor(s) affecting the R-phase transformation temperatures during post-deformation annealing, a Ti-49.8 at.% Ni and a Ti-50.8 at.% Ni alloy were subjected to various post-deformation annealing and thermal cycling treatments. The results show that the R-phase transformation temperatures are very stable in the Ti-49.8 at.% Ni alloy, while a significant variation is observed in the Ti-50.8 at.% Ni alloy with respect to the annealing and thermal cycling conditions. These findings suggest that the R-phase transformation temperatures are not susceptible to the change of dislocation density and depends mainly on the Ni concentration of the matrix, which can be modified by the formation of Ni_4Ti_3 precipitates.

Keywords: A. shape-memory alloys; B. martensitic transformation, annealing; F. differential scanning calorimetry.

1. Introduction

In near equiatomic NiTi alloys, the reversible martensitic transformation between the austenite phase (A) and the R-phase is attracting increasing interest for practical applications [1], due to its unique properties (e.g. narrow hysteresis [2,3], excellent functional fatigue resistance [4] and fast response to temperature changes [5,6]). With the presence of coherent Ni_4Ti_3 precipitates and/or dislocation networks, the R-phase can appear as an intermediate step before the formation of final B19' structured martensite (M) during cooling [7]. The coherent Ni_4Ti_3 precipitates and/or dislocation networks can be introduced by various thermomechanical treatments, such as aging treatment [3,8–10], thermal cycling [11,12] and post-deformation annealing [13–16]. The transformation temperature is a crucial factor for practical applications of the R-phase. It is therefore essential to investigate the evolution of the R-phase transformation temperatures (RTT) with respect to various thermomechanical treatments.

Coherent Ni_4Ti_3 precipitates can form in Ni-rich NiTi alloys during aging after a solution treatment, resulting in Ni depletion in the matrix [17,18] and the generation of strain fields around the precipitates [19]. Due to the small transformation strain ($\sim 1\%$) associated with $A \leftrightarrow R$ transformation [20], it is considered that the RTT in aged NiTi samples are not susceptible to the presence of the strain field, and depend mainly on the local Ni concentration in the matrix [3,10]. The R-phase transformation has been frequently observed in samples subjected to a thermally induced cyclic martensite transformation, due to the generation of dislocation networks [7,21]. With the accumulation of dislocations by increasing the thermal cycles, the volume fraction of the R-phase transformation gradually increases [22]. However, the RTT are rather stable with the increase in thermal cycles when the maximum cycling temperature (T_{\max}) is low, e.g. $T_{\max} < 423$ K [11,23]. With an elevated T_{\max} (e.g. 523 K), an obvious increase in the RTT was observed in a

Ni-rich NiTi alloy, due to the formation of Ni_4Ti_3 precipitates [12].

The effect of post-deformation annealing on the RTT is however complicated, as both the change of dislocation densities and the formation of Ni_4Ti_3 precipitates may occur. A decrease of the RTT with the increase in annealing temperatures has been frequently reported in Ni-rich NiTi alloys [13,14,24–26]. The annihilation of dislocations was considered as the main reason for the decrease of the RTT [24,27]. However, an increase of the RTT with the increase in annealing temperatures was also reported when annealing below 623 K [13,26]. Park et al. [27] suggested that the change of the RTT in Ni-rich NiTi alloys depends on the competition between the annihilation of dislocations and the formation of Ni_4Ti_3 precipitates. Todoroki and Tamura [25] found that the variation of the RTT with respect to annealing conditions becomes more pronounced with the increase in Ni content. On the other hand, the RTT are found to be very stable with respect to different annealing conditions in Ti-rich NiTi alloys [22,28]. As the annihilation of dislocations also occurs in Ti-rich NiTi alloys but not the formation of Ni_4Ti_3 precipitates [29], it seems that the change of dislocation densities during post-deformation annealing may have less effect on the RTT.

Post-deformation annealing treatment is an essential method to induce the R-phase transformation with a high transformation temperature in Ti-rich NiTi alloys, in which the formation of Ni_4Ti_3 precipitates is not occurring. Moreover, Miyazaki and Otsuka [29] reported that the reorientation of the R-phase, which is induced by the presence of dislocations networks, needs much higher stress than that of the R-phase induced by the presence of precipitates. This behavior yields interesting properties for the damping applications of the R-phase transformation. Therefore, it is important to reveal the main factor affecting the RTT during post-deformation

annealing and thus to control the RTT.

In this study, a Ti-rich and a Ni-rich NiTi alloy were subjected to various annealing treatments. With increasing annealing temperature and time, the dislocations were gradually annihilated. In the Ni-rich NiTi alloys, the formation of Ni_4Ti_3 precipitates is expected during annealing [27,29], while the influence of precipitates can be excluded in the Ti-rich NiTi alloys [29]. Besides gradually annihilating dislocations by annealing, thermal cycling tests were employed in this study to gradually accumulate dislocations. The evolution of the RTT with respect to the various treatments is investigated.

2. Materials and experimental procedures

In this study, two cold-deformed NiTi alloys with different nominal compositions (Ti-49.8 at.% Ni and Ti-50.8 at.% Ni) were used. The Ti-49.8 at.% Ni wire (diameter 7 mm) and the Ti-50.8 at.% Ni wire (diameter 2.7 mm) were produced by room-temperature wire drawing with a final pass deformation of around 20% and 35%, respectively. The as-received samples were first annealed at various temperatures between 473 and 873 K (with an interval of 50 K) for 1 hour to investigate the effect of annealing temperature on the RTT. In order to study the influence of annealing time on the RTT, the as-received samples were annealed at the temperatures between 523 and 823 K (with an interval of 100 K) for various durations from 1 to 48 hours. All the heat treatments were performed in argon atmosphere, followed by water quenching at room temperature. The transformation behavior was characterized by differential scanning calorimetry (DSC) in a TA Q2000 calorimeter between 123 and 423 K with a cooling/heating rate of 10 K min^{-1} .

The solution treated samples (annealed at 1273 K for 2 hours) were used to perform the thermal cycling tests in the same DSC instrument with a constant cooling/heating rate of 10 K min⁻¹. The transformation behavior of the solution treated samples is shown in Fig. 1. The thermal cycling tests (25 cycles) were performed between a lowest temperature (T_{\min}) of 123 K and different T_{\max} of 423 and 523 K, respectively. The samples were held at both T_{\min} and T_{\max} for 5 min during each thermal cycle.

3. Results and discussion

3.1 Effect of annealing temperatures

The transformation behavior of the as-received Ti-49.8 at.% Ni alloy after annealing at different temperatures for 1 hour is shown in Fig. 2a-b. The R-phase transformation occurs when annealing below 873 K, leading to a two-stage forward transformation $A \rightarrow R \rightarrow M$, as shown in Fig. 2a. A one-stage reverse transformation $M \rightarrow A$ is observed in all samples upon heating, as shown in Fig. 2b. The R-phase transformation start (R_s) and peak (R_p) temperatures of the Ti-49.8 at.% Ni alloy during forward transformation as a function of annealing temperature is presented in Fig. 3. Both R_s and R_p are very stable with the increase in annealing temperatures. A small scatter of R_p between 321 and 323 K with the increase in annealing temperature from 473 to 823 K is observed. This result gives a first indication that the recovery of the cold deformation sample (decrease of dislocation density) has minor influence on the RTT.

Fig. 2c-d shows the transformation behavior of the as-received Ti-50.8 at.% Ni alloy after annealing at different temperatures for 1 hour. The forward transformation occurs in two stages from austenite to the R-phase and further to martensite in the samples annealed between 523 and 823 K. A two-stage reverse transformation $M \rightarrow R \rightarrow A$ is detected in the samples annealed

between 523 and 723 K. Fig. 2c shows that the A→R transformation peak becomes sharper with the increase in annealing temperature. The reason for this phenomenon is not clear yet, and further investigations are needed. R_s and R_p of the Ti-50.8 at.% Ni alloy is plotted as a function of annealing temperature in Fig. 3. R_s and R_p show a pronounced variation, in contrast with the results of the Ti-49.8 at.% Ni alloy. R_p increases with the increase in annealing temperature from 523 to 673 K, while a decrease of R_p is observed with a further increase in annealing temperature to 823 K.

The density of dislocations gradually decreases with the increase in annealing temperature. Apart from the annihilation of dislocations, precipitation of the Ni_4Ti_3 phase is frequently observed in Ni-rich alloys at the temperature range of 473-873 K [6,10,29–32], but not in the Ti-rich alloys [29]. Therefore, by comparing the Ti-50.8 at.% Ni alloy with the Ti-49.8 at.% Ni alloy, the significant variation of R_p with respect to annealing temperatures in Ti-50.8 at.% Ni alloy must be due to the formation of Ni_4Ti_3 precipitates, which affect the Ni concentration of the matrix.

According to the equilibrium phase diagram between Ni_4Ti_3 precipitates and B2 matrix, the solid solubility of Ni in B2 matrix gradually increases with the increase in temperature [3]. The equilibrium is reached after annealing at high temperatures (723-823 K) for 1 hour, as reported in the samples subjected to aging after solution treatment [10]. Therefore, the Ni concentration in the matrix gradually increases with the increase in annealing temperatures. As a result, the RTT of Ti-50.8 at.% Ni alloy gradually decrease with the increase in annealing temperature from 723 to 823 K. However, the equilibrium cannot be reached after annealing at low temperatures (523-623 K) for only 1 hour, as reported in the samples subjected to aging after solution treatment [10]. The precipitation kinetics is mainly driven by annealing temperatures. Therefore, the volume

fraction of Ni_4Ti_3 precipitates gradually increases with the increase in annealing temperatures, leading to the increase in the Ni depletion in the matrix. As a result, the RTT of Ti-50.8 at.% Ni alloy gradually increase with the increase in annealing temperature from 523 to 623 K, as shown in Fig. 3. After annealing at 673 K for 1 hour, the Ni depletion in the matrix reaches the maximum by the formation of Ni_4Ti_3 precipitates, resulting in the highest RTT.

3.2 Effect of annealing time

Fig. 4 shows the forward transformation behavior of the Ti-49.8 at.% Ni alloy after annealing at different temperatures for 1 to 48 hours. Two-stage transformation $A \rightarrow R \rightarrow M$ is observed in the samples annealed at 523 to 723 K for different times, as shown in Fig. 4a-c. The $A \rightarrow R$ and $R \rightarrow M$ transformation peaks overlap with each other when annealing at 823 K for 1 hour, as shown in Fig. 4d. Fig. 5 shows the variation of R_p of the Ti-49.8 at.% Ni alloy with respect to annealing time at different annealing temperatures. R_p lies in a narrow temperature range between 321 and 324 K. As the dislocation density gradually decreases with the increase in annealing time or temperature, the small scatter of R_p indicates that the RTT are not susceptible to the change of dislocation densities.

The forward transformation behavior of the Ti-50.8 at.% Ni alloy after annealing at different temperatures for 1 to 48 hours is shown in Fig. 6. The R-phase transformation is observed in the samples annealed at 523 to 723 K for different times, as presented in Fig. 6a-c. Two-stage transformation $A \rightarrow R \rightarrow M$ is observed in the sample annealed at 823 K for 1 hour, while with the increase of annealing time to 4 and 8 hours, multi-stage transformation is detected, as shown in Fig. 6d. The severe decrease of dislocations and heterogeneous distribution of the Ni_4Ti_3 precipitates are considered as the main reason for the appearance of the multi-stage

transformation, as reported in aged samples [33,34]. Nevertheless, further investigation is required to explain this phenomenon.

Fig. 7 shows the variation of R_p of the Ti-50.8 at.% Ni alloy with respect to annealing time at different annealing temperatures. R_p continuously increases from 273 K to 308 K with increasing annealing time from 1 to 48 hours when annealing at 523 K. The increase of R_p is attributed to the continuous Ni depletion in the matrix, which is caused by the increase in the volume fraction of Ni_4Ti_3 precipitates with increasing annealing time. When annealing at 623 K, R_p first increases from 306 to 317 K with increasing annealing time from 1 to 8 hours. R_p remains stable with further increasing annealing time to 24 and 48 hours, as shown in Fig. 7. The increase of R_p is also due to the increasing volume fraction of Ni_4Ti_3 precipitates with increasing annealing time. The equilibrium condition between the B2 matrix and Ni_4Ti_3 precipitate can be reached after annealing for 24 hours. The Ni concentration maintains the equilibrium value, thus R_p remains stable with increasing annealing time from 24 to 48 hours.

R_p varies between 303 and 308 K with increasing annealing time from 1 to 48 hours when annealing at 723 K. This insensitivity of R_p with respect to the annealing time is due to the fact that the equilibrium condition between Ni_4Ti_3 precipitates and the B2 matrix is already achieved after annealing at 723 K for 1 hour. Xue et al. [10] reported the same phenomenon in the samples subjected to aging after solution treatment. The Ni concentration of the matrix maintains the equilibrium value, thus R_p remains stable with increasing annealing time from 1 to 48 hours. R_p with a value of 260 K is observed in the sample annealed at 823 K for 1 hour. This is due to the high solubility of Ni in the matrix at the temperature of 823 K, as indicated by the equilibrium phase diagram between Ni_4Ti_3 and B2 matrix [3].

3.3 Effect of thermal cycling

Dislocations can be gradually accumulated with the increase in the number of thermally induced martensite transformation cycles [11,35]. Moreover, with elevated maximum cycling temperature (T_{\max}), Ni_4Ti_3 precipitates can be introduced into the Ni-rich alloys during thermal cycling [12]. As shown in Fig. 8, the Ti-49.8 at.% Ni and Ti-50.8 at.% Ni alloys are subjected to thermal cycling tests with different T_{\max} , in order to separate the influence of the change in dislocation densities and the formation of Ni_4Ti_3 precipitates on the RTT.

Fig. 8a-b shows the evolution of the transformation behavior of the Ti-49.8 at.% Ni alloy during thermal cycling between T_{\min} of 123 K and different T_{\max} of 423 and 523 K, respectively. Only the DSC data between 150 and 400 K is presented, as the heat flow remains constant outside of this temperature range. The samples were isothermally held at both T_{\min} and T_{\max} for 5 min during each cycle. As shown in Fig. 8a, A \rightarrow R transformation is observed in the 7th cycle for the test with T_{\max} of 423 K. With a higher T_{\max} of 523 K, A \rightarrow R transformation can be detected from the 12th cycle on, as shown in Fig. 8b. This is due to the slight annihilation of dislocations at the temperature of 523 K, which delays the accumulation of dislocations during thermal cycling. In both samples, the B19' martensite transformation is gradually suppressed with the increase in thermal cycles.

The transformation behavior of the Ti-50.8 at.% Ni alloy during thermal cycling between T_{\min} of 123 K and different T_{\max} of 423 and 523 K are respectively shown in Fig. 8c and 8d. A distinguishable R-phase transformation peak is observed in the 17th cycle for the test with T_{\max} of 423 K, as shown in Fig. 8c. Comparing with the Ti-49.8 at.% Ni alloy (Fig. 8a), the occurrence of the R-phase transformation is much later in the Ti-50.8 at.% Ni alloy. This is due to the fact that

the excess solute Ni atoms hinder the accumulation of dislocations in the Ti-50.8 at.% Ni alloy [11,36]. The R-phase transformation occurs much earlier in the Ti-50.8 at.% Ni alloy when subjected to the thermal cycling with T_{\max} of 523 K, despite of the slight annihilation of dislocations. Moreover, the suppression of martensite transformation is pronounced in the sample subjected to thermal cycling with T_{\max} of 523 K.

Fig. 9a shows the evolution of R_p and M_p of the Ti-49.8 at.% Ni alloy during thermal cycling tests. R_p of the Ti-49.8 at.% Ni alloy is constant during thermal cycling with T_{\max} of both 423 and 523 K. R_p maintains at 317 K during the thermal cycling tests with T_{\max} of 423 K, while R_p becomes 320 K for the test with T_{\max} of 523 K. The difference of R_p between the test with T_{\max} of 423 and 523 K is attributed to the slight composition variation between the samples used in the tests. With the increase in the number of thermal cycles, M_p gradually decreases for both samples, due to the accumulation of dislocations [11], which act as the obstacles for martensite transformation [3,7]. Due to the slight annihilation of dislocations with a higher T_{\max} of 523 K, the accumulation of dislocations is delayed. As a result, martensite transformation is less suppressed in the sample subjected to the thermal cycling test with T_{\max} of 523 K.

The evolution of R_p and M_p of the Ti-50.8 at.% Ni alloy during thermal cycling tests is shown in Fig. 9b. With T_{\max} of 423 K, the thermal cycling hardly affects the R-phase transformation temperature, and R_p maintains at around 234 K. As the temperature is low, almost no Ni_4Ti_3 precipitates are expected during the thermal cycling test with T_{\max} of 423 K. With the increase of T_{\max} to 523 K, R_p gradually increases from 242 to 280 K with increasing number of thermal cycles from 6 to 25. The increase of the RTT is attributed to the continuous Ni depletion by increasing the volume fraction of Ni_4Ti_3 precipitates with increasing number of thermal cycles

[12]. As shown in Fig. 9b, a pronounced decrease of M_p from 232 to 177 K is observed in the Ti-50.8 at.% Ni alloy when subjected to thermal cycling with T_{max} of 523 K, despite of the slight annihilation of dislocations. This is attributed to the formation of coherent Ni_4Ti_3 nanoprecipitates during the thermal cycling with T_{max} of 523 K [12], which suppresses the martensite transformation [6,7,37].

The above findings show that the evolution of the RTT during thermal cycling depends on the composition of the samples as well as T_{max} . When no Ni_4Ti_3 precipitates are introduced (e.g., in the Ti-49.8 at.% Ni alloy or in the Ti-50.8 at.% Ni alloy subjected to thermal cycling with T_{max} of 423 K), the RTT are stable even though the dislocations are gradually accumulated with an increase in the number of thermal cycles. Due to the gradual Ni depletion in the matrix, which is caused by the formation of Ni_4Ti_3 precipitates, the increase in the RTT is observed in Ti-50.8 at.% Ni alloy during thermal cycling with an elevated T_{max} of 523 K.

4. Conclusions

In this study, cold deformed Ti-49.8 at.% Ni and Ti-50.8 at.% Ni alloys were subjected to post-deformation annealing and thermal cycling treatments under various conditions. The evolution of the R-phase transformation temperatures has been investigated. The results observed in this study has led us to conclude that the R-phase transformation temperature is not susceptible to the density of dislocations and depends mainly on the Ni concentration of the matrix, which can be modified by the presence of Ni_4Ti_3 precipitates.

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Figures:

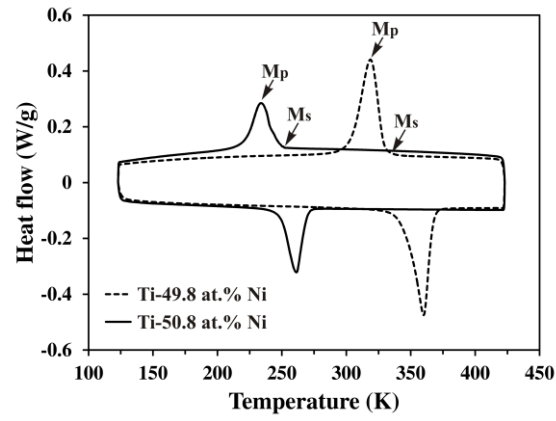


Fig. 1. Transformation behavior of the Ti-49.8 at.% Ni and Ti-50.8 at.% Ni alloy after annealing at 1273 K for 2 hours.

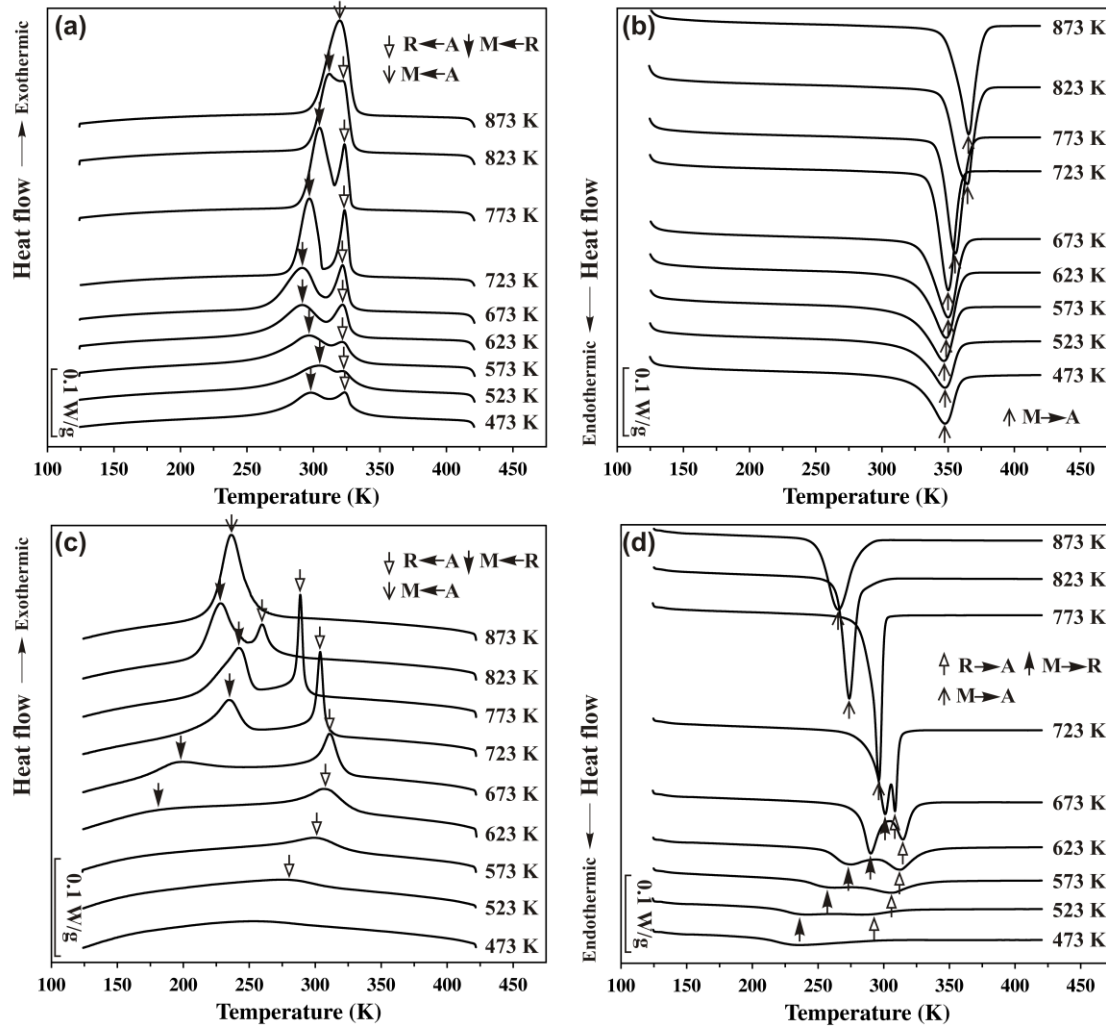


Fig. 2. Transformation behavior of the as-received Ti-49.8 at.% Ni and Ti-50.8 at.% Ni alloy after annealing at different temperatures for 1 hour. (a) Forward and (b) reverse transformation of the Ti-49.8 at.% Ni alloy; (c) forward and (d) reverse transformation of the Ti-50.8 at.% Ni alloy.

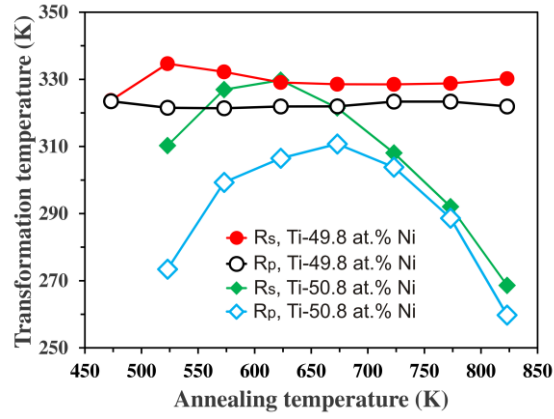


Fig. 3. The evolution of the R-phase transformation start (R_s) and peak (R_p) temperature of the Ti-49.8 at.% Ni and Ti-50.8 at.% Ni alloy with respect to annealing temperature.

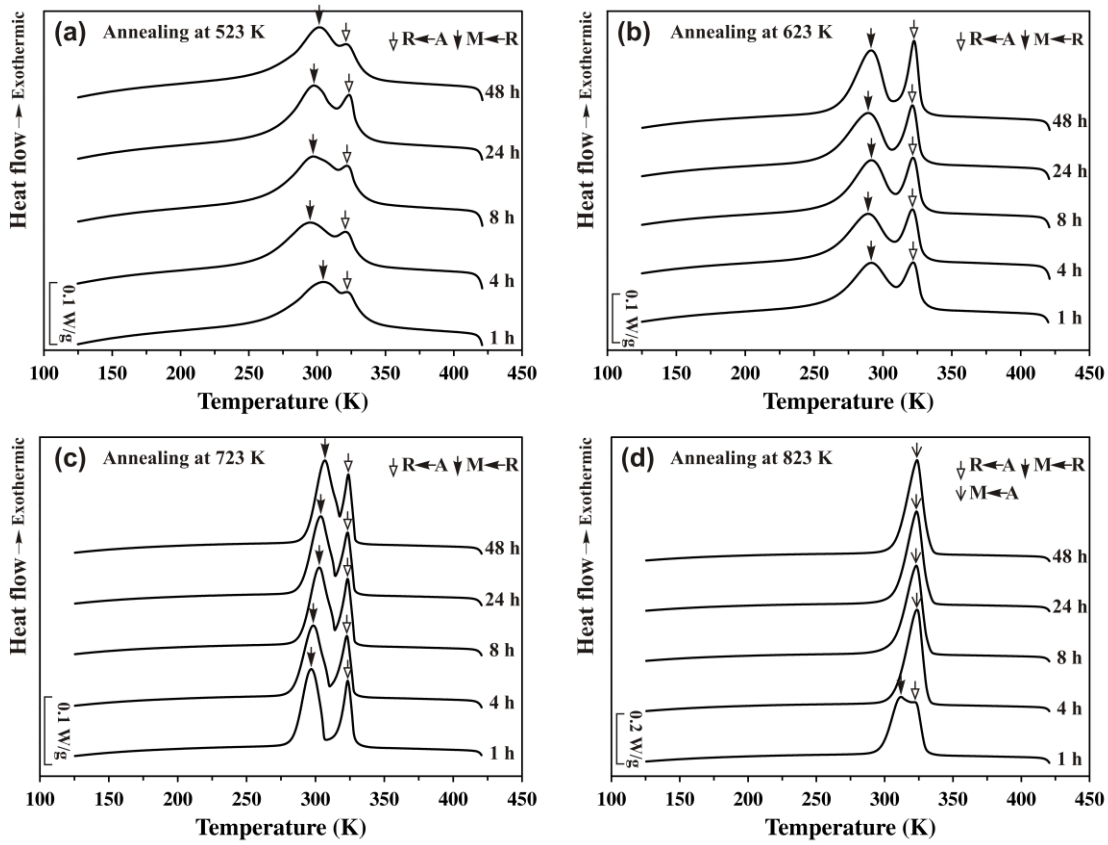


Fig. 4. Transformation behavior of the Ti-49.8 at.% Ni alloy after annealing at (a) 523 K; (b) 623 K; (c) 723 K and (d) 823 K for different durations.

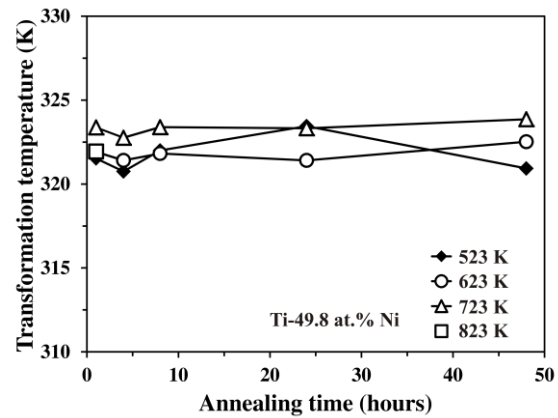


Fig. 5. The R-phase transformation peak temperature (R_p) of the Ti-49.8 at.% Ni alloy as a function of annealing time when annealed at different temperatures.

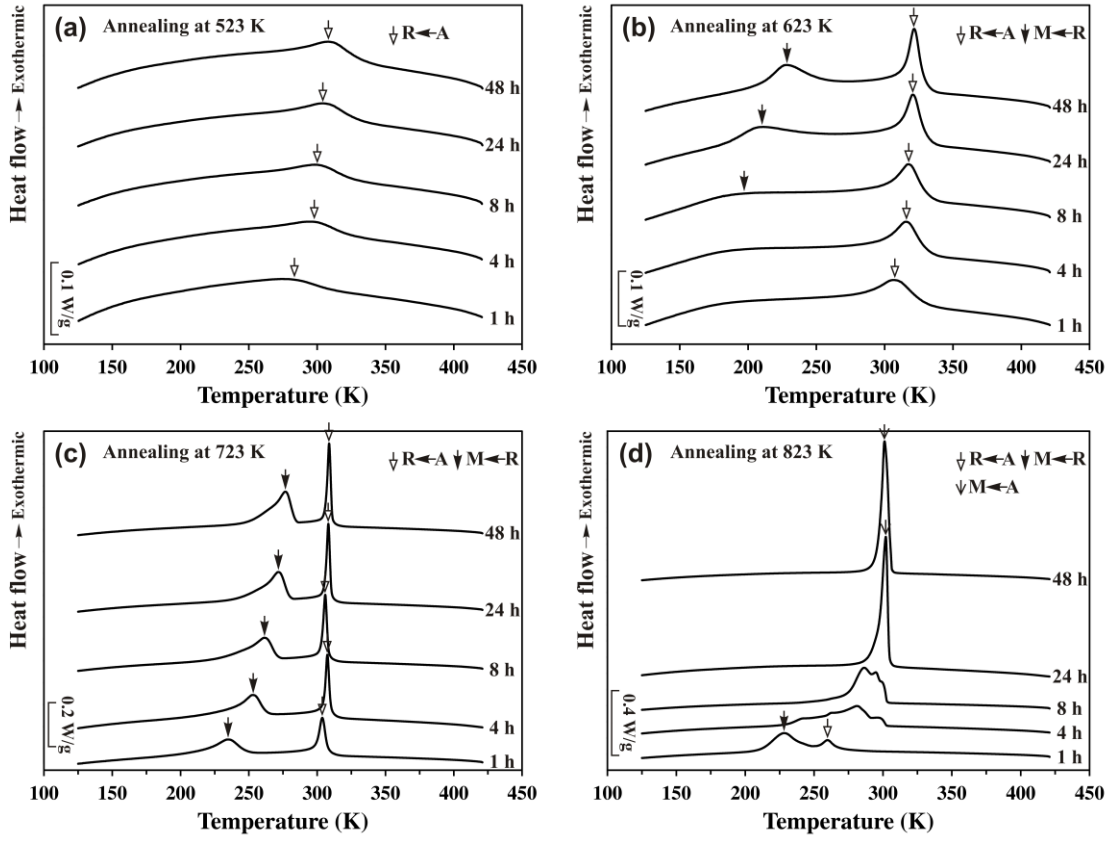


Fig. 6. Transformation behavior of the Ti-50.8 at.% Ni alloy after annealing at (a) 523 K; (b) 623 K; (c) 723 K and (d) 823 K for different durations.

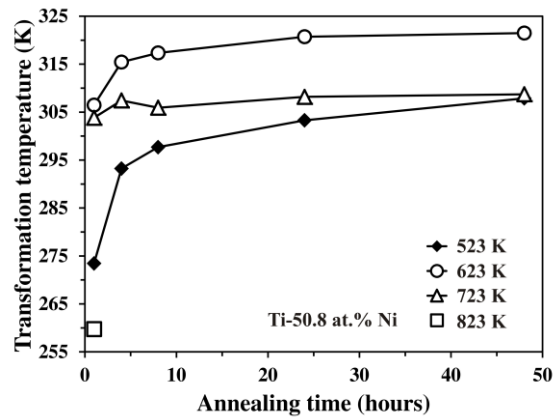


Fig. 7. The R-phase transformation peak temperature (R_p) of the Ti-50.8 at.% Ni alloy as a function of annealing time when annealed at different temperatures.

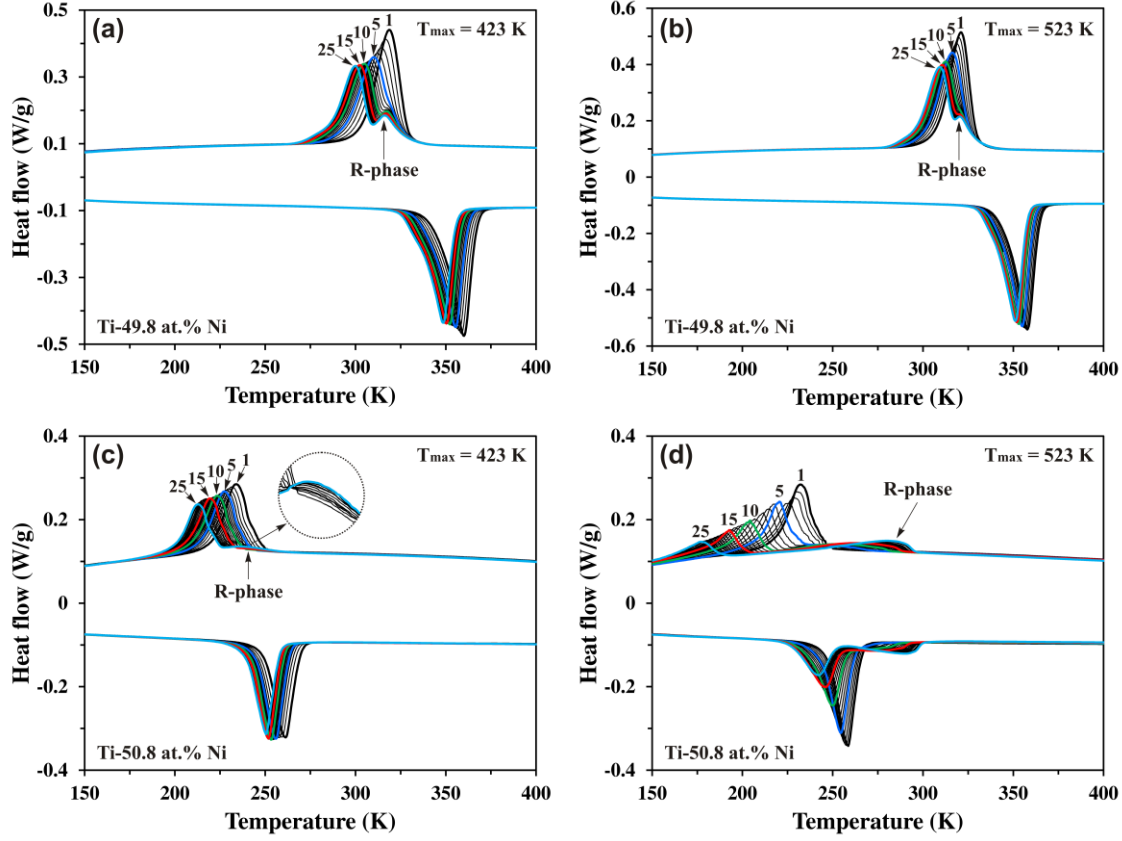


Fig. 8. Transformation behavior during thermal cycling between 123 K and different maximum temperatures (T_{\max}): (a) Ti-49.8 at.% Ni alloy with T_{\max} of 423 K; (b) Ti-49.8 at.% Ni alloy with T_{\max} of 523 K; (c) Ti-50.8 at.% Ni alloy with T_{\max} of 423 K; and (d) Ti-50.8 at.% Ni alloy with T_{\max} of 523 K. The samples were annealing at 1273 K for 2 hours before thermal cycling tests.

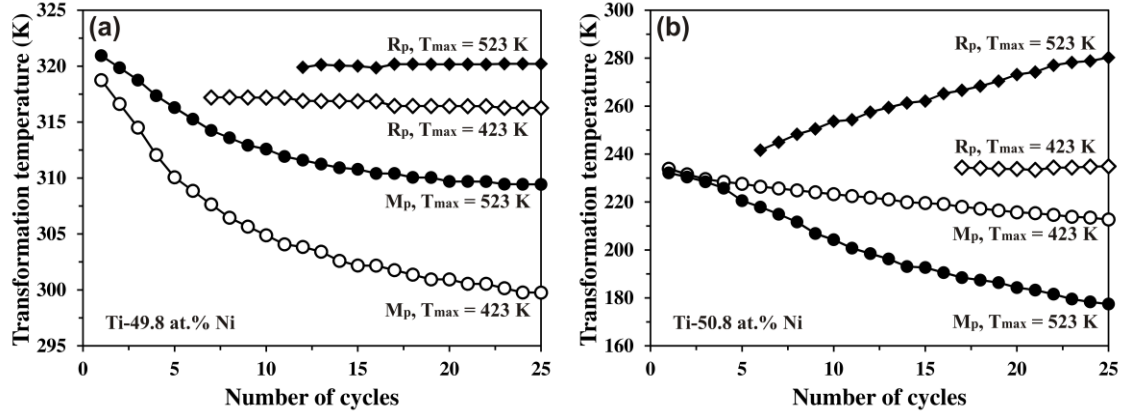


Fig. 9. Evolution of the peak transformation temperature of the R-phase (R_p) and martensite (M_p) transformation during thermal cycling with different T_{max} : (a) Ti-49.8 at.% Ni alloy and (b) Ti-50.8 at.% Ni alloy.